

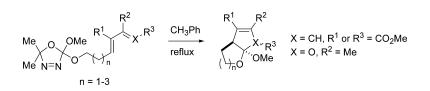
Communication

Inter- and Intramolecular [4 + 1]-Cycloadditions Between Electron-Poor Dienes and Electron-Rich Carbenes

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Inter- and Intramolecular [4 + 1]-Cycloadditions Between Electron-Poor Dienes and Electron-Rich Carbenes

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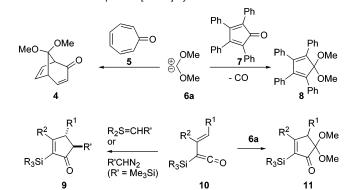
The [4 + 1]-cycloaddition between a carbene **2** and a diene **1** to give a cyclopentene derivative is, in a sense, the five-membered ring equivalent of the Diels-Alder reaction (eq 1).



Given the ubiquity of five-membered carbo- and heterocyclic substructures in natural products, one would expect the [4 + 1]-cycloaddition, like the Diels–Alder reaction, to count among the most powerful reactions in the arsenal of the synthetic chemist. On the contrary, this reaction has few precedents. Lilienblum and Hoffmann reported such an intermolecular cheletropic reaction between dimethoxycarbene **6a** and tropone **5** or tetracyclone **7** (Scheme 1).¹ The same carbene also added to tetrazines² and to 4,4-bis(trifluoromethyl)-1,3-diazabutadiene.³ More recently, Rigby established that **6a** added to vinyl isocyanates to give five-membered ring lactams.⁴ In addition, dimethoxy carbene **6a** reacts with silyl-substituted vinyl ketenes **10** (Scheme 1).⁵ The latter also react with sulfur ylids or trimethylsilyldiazomethane to give **9**.⁶

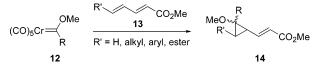
Examples of [4 + 1]-cycloadditions between a carbene and a simple diene are rare.⁷ The main reason is the known propensity of carbenes and carbenoids to give cyclopropanation products with alkenes and 1,3-dienes.⁸ For example, dimethoxycarbene gave cyclopropane derivatives in low yield with 1-phenyl- and 1,1diphenyl-1,3-butadiene.¹ Fisher carbene complexes **12** gave mostly cyclopropanation products with electron-poor dienes (Scheme 2).⁹ On the basis of the precedented rearrangement of vinylcyclopropanes,¹⁰ Hudlicky¹¹ and Danheiser¹² established a two-step protocol that converted 1,3-dienes to cyclopentene derivatives.

Warkentin and co-workers studied extensively the chemistry and mechanistic aspects of dialkoxy carbenes and related bis-heteroatom carbenes.¹³ One of the easiest methods to generate such carbenes is through the thermal decomposition of appropriately substituted oxadiazolines 15 (Scheme 3).^{13,14} We were surprised to find that dialkoxycarbenes 6a and 6b added efficiently to electron-poor diene 17 to give the corresponding cyclopentene adducts 18a and 18b in 67 and 60% yields, respectively. Cyclopropane products were not detected, although they could be intermediates (vide infra). Dienes 19a,b gave cyclopentanone acetals 20a,b as single diastereomers in 38% and 29% yield, respectively. The latter reaction gave a crude product with a clean NMR that indicated complete conversion of the diene 19b, but the products were somewhat unstable. The two methyl ether groups in 20a gave different ¹H NMR signals, indicating a cis stereochemistry of the two esters in 20a. The stereochemistry of 20b was assumed to be the same as 20a. Diene 21 reacted efficiently also, as judged by NMR, but gave cyclopentadiene 22 in 30% yield after purification on silica gel. Vinylcyclohexene and other electron-rich dienes, such as Danishefsky's diene, did not react with 6a.

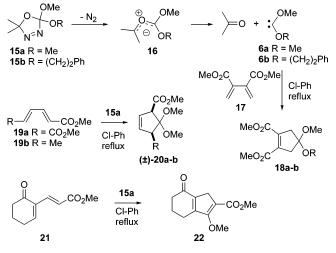


Scheme 1. Examples of [4 + 1]-cycloadditions

Scheme 2. Fisher Carbenes 12 Give Mostly Cyclopropane Products with Electron-Poor Dienes

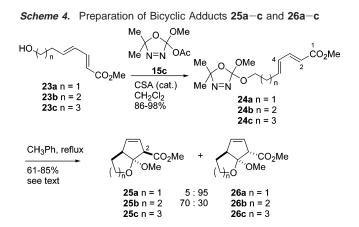


Scheme 3. Intermolecular [4 + 1]-Cycloadditions of Electron-Deficient Dienes with Dialkoxy Carbenes

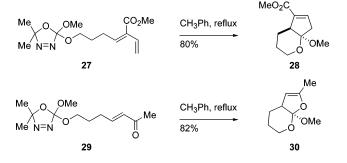


At this point, we were convinced that an intramolecular version of this annulation reaction could lead to useful polycyclic structures in improved yields. We are not aware of prior examples of such intramolecular [4 + 1]-cycloaddition aside from a fortuitous reaction observed by Warkentin and co-workers.¹⁵

We prepared dienes 23a-c, which were coupled to oxadiazoline 15c using an improved reaction condition (Scheme 4).¹⁶ Heating to reflux the resulting oxadiazolines 24a-c in toluene in carefully washed glassware gave good yields of the bicyclic adducts 25a-c and 26a-c (Scheme 4). Oxadiazoline 24a gave a 5:95 mixture of diastereometric adducts 25a and 26a in 85% yield, while acetals



Scheme 5. Preparation of Bicyclic Acetal 28 and Orthoester 30



25b and 26b were isolated in a ratio of 70:30 in 61% combined yield. Reduction of the ester in 26a, 25b, and 26b to the primary alcohol and subsequent derivatization into the corresponding *p*-nitrobenzoate ester allowed assignment of their stereochemistry on the basis of single-crystal X-ray diffraction analyses. Oxadiazoline 24c led to a single diastereomer, although we could not yet unambiguously assign its stereochemistry.

Heating 27 gave diastereomerically pure 28 in 80% yield along with 16% of a carbonate, a frequent byproduct of the thermolysis of oxadiazolines in general (Scheme 5).¹³ We were also pleased to see that [4 + 1]-annulation was possible with enone 29 to give a bicyclic ortho ester 30 in 82% isolated yield.

The mechanism of this annulation process is not clear at present. Carbene 6a is nucleophilic and generally reacts with electrophiles to form zwitterionic intermediates.^{4a,13} Our results seem to indicate that this may not always be the case. While it is tempting to assume a general stepwise mechanism for these [4 + 1]-annulation reactions on the basis of the formation of both isomers 25 and 26, the isolation of 20a,b as single isomers is indicative of a concerted process. In that regard, we have prepared the (2Z,4E)-isomer of **24a**, and it does not give any observable amount of 25a or 26a under the same reaction conditions as per (E,E)-24a. Instead, several decomposition products were reproducibly obtained. We would expect a stepwise process to be independent of the geometry of the C2-double bond in 24a. Moreover, it is difficult to explain the reversal in selectivity in the cycloadditions of the carbenes derived from 24a and 24b, even if the cyclizations were entirely reversible. Further probing into the mechanism of this reaction is ongoing in our laboratories.

In conclusion, we have shown that electron poor dienes and oxadienes participate in [4 + 1]-cycloadditions with dialkoxy carbenes. The intramolecular reactions are the first examples of this kind and lead to useful bicyclic structures. We are presently expanding this methodology to include nitrogen^{17,18} and sulfur^{5,19,20} atoms as well as probing into the mechanistic aspects of the reaction.

Acknowledgment. We thank the Natural Sciences and Engineering Council of Canada and the Université de Sherbrooke for financial support.

Supporting Information Available: Experimental and ¹H NMR spectra of all new compounds and syntheses and characterization of 23a-c (PDF, CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) Lilienblum, W.; Hoffman, R. W. Chem. Ber. 1977, 110, 3405-3409. (2) Gerninghaus, C.; Kümmell, A.; Seitz, G. Chem. Ber. 1993, 126, 733-738
- (3) Burger, K.; Wassmuth, U.; Penninger, S. J. Fluor. Chem. 1982, 20, 813-825
- (4)(a) Rigby, J. H.; Cavezza, A.; Ahmed, G. J. Am. Chem. Soc. 1996, 118, 12848–12849. (b) Rigby, J. H.; Cavezza, A.; Heeg, M. J. J. Am. Chem. Soc. 1998, 120, 3664–3670.
- Rigby, J. H.; Wang, Z. Org. Lett. 2003, 5, 263–264. Loebach, J. L.; Bennett, D. M.; Danheiser, R. L J. Am. Chem. Soc. 1998, (6)120, 9690-9691.
- Exceptional cases include: (a) Davies, H. M. L.; Clark, T. J. *Tetrahedron* **1994**, 50, 9883–9892. (b) Schnaubelt, J.; Marks, E.; Reissig, H.-U. *Chem. Ber.* **1996**, 129, 73–75. Special dienes: (c) Burger, U.; Gandillon, G.
 Tetrahedron Lett. **1979**, 20, 4281–4284. (d) Mayr, H.; Heigl, U. W. (7)Angew. Chem., Int. Ed. Engl. 1985, 24, 579-580. (e) Sugita, H.; Mizuno, K.; Mori, T.; Isagawa, K.; Otsuji, Y. Angew. Chem., Int. Ed. Engl. 1991, 30, 984-986
- Moss, R. A.; Jones, M., Jr. In Reactive Intermediates; Wiley: New York, 1981; Vol. 2, Chapter 3, and references therein
- (a) Buchert, M.; Hoffmann, M.; Reissig, H.-U. Chem. Ber. 1995, 128, 605–614. (b) Buchert, M.; Reissig, H.-U. Chem. Ber. 1992, 125, 2723– (9)2729. (c) Buchert, M.; Reissig, H.-U. Tetrahedron Lett. 1988, 29, 2319-2320.
- (10) For reviews, see: (a) Frey, H. M.; Walsh, R. Chem. Rev. 1969, 69, 103-124. (b) Sarel, S.; Yovell, J.; Sarel-Imber, M. Angew. Chem., Int. Ed. Engl. 1968, 7, 577–588. See also: (c) Mazzocchi, P. H.; Tamburin, H. J. J. Am. Chem. Soc. 1975, 97, 555–561. (d) Willcott, M. R., III; Cargle, V. H. J. Am. Chem. Soc. 1969, 91, 4310-4311. (e) Willcott, M. R., III; Cargle, V. H. J. Am. Chem. Soc. 1967, 89, 723-724.
- (a) Hudlicky, T.; Koszyk, F. J.; Kutchan, T. M.; Sheth, J. P. J. Org. Chem. **1980**, 45, 5020–5027. (b) Hudlicky, T.; Koszyk, F. J.; Dochwat, D. M.; Cantrell, G. L. J. Org. Chem. **1981**, 46, 2911–2915. (c) Short, R. P.; (11)Revol, J.-M.; Ranu, B. C.; Hudlicky, T. J. Org. Chem. 1983, 48, 4453-4461
- (12) (a) Danheiser, R. L.; Bronson, J. J.; Okano, K. J. Am. Chem. Soc. 1985, 107, 4579–4582. (b) Danheiser, R. L.; Martinez-Davilla, C.; Auchus, R. J.; Kadonaga, J. T. J. Am. Chem. Soc. **1981**, 103, 2443–2446. (c) Danheiser, R. L.; Martinez-Davilla, C.; Morin, J. M., Jr. J. Org. Chem. 1980, 45, 1340-1341.
- (13) For a review of his many contributions, see: Warkentin, J. J. Chem. Soc., Perkin Trans. 1 2000, 2161–2169 and references therein.
- (14) El-Saidi, M.; Kassam, K.; Pole, D. L.; Warkentin, J. J. Am. Chem. Soc. 1992, 114, 8751–8752.
- (15) Lu, X.; Warkentin, J. Tetrahedron Lett. 1999, 40, 1483-1486.
- (16) (a) Kassam, K.; Warkentin, J. Can. J. Chem. 1997, 75, 120-128. (b) Lu, X.; Reid, D. L.; Warkentin, J. Can. J. Chem. 2001, 79, 319-32
- (17) (a) Rigby, J. H.; Wang, Z. Org. Lett. 2002, 4, 4289-4291. (b) Rigby, J. H.; Cavezza, A.; Heeg, M. J. Tetrahedron Lett. 1999, 40, 2473-
- (18) (a) Couture, P.; Terlouw, J. K.; Warkentin, J. J. Am. Chem. Soc. 1996, 118, 4214–4215. (b) Couture, P.; Warkentin, J. Can. J. Chem. 1997, 75, 1264-1280. (c) Couture, P.; Warkentin, J. Can. J. Chem. 1997, 75, 1281 1294. (c)
- (a) Rigby, J. H.; Laurent, S. J. Org. Chem. **1999**, 64, 1766–1767. (b) Rigby, J. H.; Dong, W. Org. Lett. **2000**, 2, 1673–1675. (c) Rigby, J. H.; (19)Danca, D. Tetrahedron Lett. **1999**, 40, 6891–6894.
- (20) Er, H.-T.; Pole, D. L.; Warkentin, J. Can. J. Chem. 1996, 74, 1480-1489. JA046344X